Data collection	
AFC-6S diffractometer	$\theta_{\rm max} = 24.97^{\circ}$
$\omega/2\theta$ scans	$h = 0 \rightarrow 10$
Absorption correction:	$k = 0 \rightarrow 17$
not applied	$l = -16 \rightarrow 16$
3649 measured reflections	3 standard reflections
3649 independent reflections	monitored every 15
1786 observed reflections	reflections
$[I>3\sigma(I)]$	intensity variation:
Refinement	
Refinement on F	$(\Delta/\sigma)_{\rm max} = 0.0003$
R = 0.0443	$\Delta \rho_{\rm max} = 0.23 \ {\rm e} \ {\rm \AA}^{-3}$
wR = 0.0494	$\Delta \rho_{\rm min} = -0.23 \ {\rm e} \ {\rm \AA}^{-3}$
S = 1.627	Extinction correction:
1786 reflections	applied
291 parameters	Atomic scattering fact

All H-atom parameters refined Weighting scheme based on measured e.s.d.'s

0 none

not Atomic scattering factors from International Tables for X-ray Crystallography (1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ($Å^2$)

$U_{\text{eq}} = (1/3) \sum_i \sum_j U_{ij} a_i^* a_i^* \mathbf{a}_i \cdot \mathbf{a}_j.$

	x	у	z	U_{eq}
Cll	1.0108(1)	0.47234 (10)	0.28143 (10)	0.0589
Cl7	0.3397 (1)	0.35103 (9)	0.21595 (9)	0.0500
CI14	0.6096 (2)	0.20949 (7)	0.23810 (9)	0.0539
N13	0.7972 (4)	0.3454 (2)	0.2547 (2)	0.0360
C1	0.8075 (5)	0.4431 (3)	0.2621 (3)	0.0365
C2	0.7390 (5)	0.4873 (3)	0.1681 (3)	0.0355
C3	0.8260 (6)	0.4707 (4)	0.0758 (4)	0.0547
C4	0.7207 (6)	0.4149 (3)	0.0115 (3)	0.0466
C5	0.5796 (5)	0.4009 (3)	0.0499 (3)	0.0388
C6	0.5716 (5)	0.4498 (3)	0.1463 (3)	0.0308
C7	0.5320 (5)	0.3919 (3)	0.2352 (3)	0.0334
C8	0.5512 (5)	0.4479 (3)	0.3320 (3)	0.0318
C9	0.5184 (6)	0.3960 (3)	0.4227 (3)	0.0401
C10	0.6513 (6)	0.3866 (3)	0.4854 (3)	0.0432
C11	0.7873 (7)	0.4331 (4)	0.4481 (3)	0.0483
C12	0.7256 (5)	0.4745 (3)	0.3490 (3)	0.0349
C14	0.6566 (5)	0.3217 (3)	0.2439 (3)	0.0324
C15	0.4664 (7)	0.3501 (3)	-0.0020 (4)	0.0526
C16	0.503(1)	0.3127 (4)	-0.0914 (4)	0.0692
C17	0.6441 (9)	0.3290 (4)	-0.1280 (4)	0.0697
C18	0.7529 (8)	0.3793 (4)	-0.0776 (4)	0.0664
C19	0.6473 (8)	0.3392 (4)	0.5724 (4)	0.0615
C20	0.5093 (9)	0.3030(4)	0.5955 (4)	0.0683
C21	0.3763 (9)	0.3146 (4)	0.5364 (4)	0.0674
C22	0.3787 (7)	0.3624 (4)	0.4493 (4)	0.0533

Table 2.	Selected	geometric	parameters	(A,	°)	I
		A	p	· 7		

	Ų	-	• • •
Cl1-Cl	1.809 (4)	C1-C12	1.507 (6)
Cl7—C7	1.771 (4)	C2C6	1.559 (6)
Cl14-C14	1.729 (4)	C6—C7	1.557 (6)
N13-C1	1.468 (6)	C7—C8	1.568 (6)
N13-C14	1.263 (6)	C7—C14	1.503 (6)
C1C2	1.528 (6)	C8-C12	1.558 (6)
N13-C1-C2	110.9 (3)	C6-C7-C14	104.6 (3)
N13-C1-C12	109.6 (4)	C8-C7-C14	106.2 (3)
C2-C1-C12	110.9 (4)	Cl14-C14-N13	120.0 (3)
C6-C7-C8	110.5 (3)	Cl14-C14-C7	120.8 (3)

Data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1988). Cell refinement: MSC/AFC Diffractometer Control Software. Data reduction: TEXSAN PROCESS (Molecular Structure Corporation, 1985). Program(s) used to solve structure: TEXSAN, MITHRIL (Gilmore, 1984). Program(s) used to refine structure: TEXSAN LS. Molecular graphics: ORTEPII (Johnson, 1976). Software used to prepare material for publication: TEXSAN FINISH. Literature survey: CSSR (1984).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71722 (27 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England, [CIF reference: HU1062]

References

- Barlow, M. G., Haszeldine, R. N. & Simpkin, D. J. (1982). J. Chem. Soc. Perkin Trans. 1, pp. 1245-1249.
- CSSR (1984). Crystal Structure Search and Retrieval Instruction Manual. SERC Daresbury Laboratory, Warrington, England.
- Gilmore, C. J. (1984). J. Appl. Cryst. 17, 42-46.
- Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Molecular Structure Corporation (1985). TEXSAN. TEXRAY Structure Analysis Package. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Molecular Structure Corporation (1988). MSC/AFC Diffractometer Control Software. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

Acta Cryst. (1994). C50, 553-556

exo.exo-4,4,12,12,16,16-Hexakis(trifluoromethyl)-17-(3,3,3-trifluoro-2-trifluoromethyl-1-azapropenyl)-3,11,17-triazaheptacyclo[12.4.1.1^{6,9}.0^{2,13}.0^{3,11}.0^{5,10}.0^{15,18}]icos-7-ene Formed via Novel 1,3-Dipolar Cycloaddition to Quadricyclane

M. G. BARLOW, R. G. PRITCHARD, N. S. SULIMAN AND A. E. TIPPING

Department of Chemistry, University of Manchester Institute of Science and Technology, PO Box 88, Manchester M60 1QD, England

(Received 30 April 1993; accepted 24 August 1993)

Abstract

The title compound, $C_{26}H_{16}F_{24}N_4$, crystallizes as a racemic mixture with two crystallographically independent molecules in the asymmetric unit, which differ only slightly in conformation. In both cases, the central diazo region bears a close resemblence to the structure of diethyl *trans*-4,4,8,8-tetrakis(trifluoromethyl)-1,5diazabicyclo[3.3.0]octane-2,6-dicarboxylate [Burger, Schickaneder, Hein, Gieren, Lamm & Engelhardt (1992). *Liebigs Ann. Chem.* pp. 845-852], in which the N—N bond is bisected by a crystallographic inversion centre.



Fig. 1. The two crystallographically independent molecules of the title compound, including atomic numbering scheme, drawn using *OR*-*TEPII* (Johnson, 1976). The F and H labels are based on those of the bonded C atoms.

Comment

The structure determination reported herein forms the basis of an investigation of novel 1,3-dipolar cycloadditions of azines and azomethine imines to quadricyclanes (Barlow, Suliman & Tipping, 1992). Cycloaddition $(\sigma^2 s + \sigma^2 s + \pi^2 s)$ of hexafluoroacetone azine (HFAA) to quadricyclane gives the 1:1 adduct (2). This can then undergo further cycloaddition with either (i) quadricyclane $(\sigma^2 s + \sigma^2 s + \pi^2 s)$ to give the 1:2 adduct (3) or (ii) the azine $(4\pi + 2\pi)$ to afford the 2:1 adduct (4). 1,3-Dipolar cycloaddition of the 2:1 adduct (4) to quadricyclane by a two-step mechanism (a concerted $2s + 2s + 4\pi$ addition is disallowed thermally) then gives the title compound, the 2:2 adduct (1).



Experimental

A mixture of quadricyclane (1.80 g, 19.56 mmol), hexafluoroacetone azine (3.50 g, 10.60 mmol) and dichloromethane (2 ml) was sealed in vacuo in a Rotaflo tube (ca 50 ml) and heated at 343 K for 2 d. The resulting material was washed from the tube with dichloromethane (10 ml) and the solvent removed (rotary evaporator), thus producing a mixture (4.22 g) of three products as shown by TLC (R_F 0.88, 0.74 and 0.65; eluant hexane). Chromatographic separation (Merck Kieselgel 60; eluant hexane) gave: (i) exo, exo-4, 4-bis(trifluoromethyl)-3-{4,4bis(trifluoromethyl)-3-azatricyclo[4.2.1.0^{2,5}]non-7-en-3-yl}-3azatricyclo[4.2.1.0^{2,5}]non-7-ene (3) (0.21 g, 0.41 mmol, 8%; found M⁺ 512.1115; C₂₀H₁₆F₁₂N₂ requires M 512.1008), m.p. 385-386 K; (ii) exo-4,4-bis(trifluoromethyl)-3-(3,3,3-trifluoro-2-trifluoromethyl-1-azapropenyl)-3-azatricyclo[4.2.1.0^{2,5}]non-7-ene (2) (2.40 g, 5.71 mmol, 54%; found C 37.2, H 1.9, N 6.9, F 54.8, M⁺ 420.0486; C₁₃H₈F₁₂N₂ requires 37.1, H 1.9, N 6.7, F 54.3, M 420.0496), b.p. 392-С

C 37.1, H 1.9, N 6.7, F 54.3, M 420.0496), b.p. 392-394 K; and (iii) the title compound exo, exo-4, 4, 12, 12, 16, 16-

hexakis(trifluoromet	hvl)_17_(3.3.3	trifluoro 7 trifluo	romothul	E26 P	0.0154 (2)	0.0226 (2)	0.00001 (10)	
1_{-373} horses 1_{-373} 1_{-17} 1_{-1			F26C	0.9134 (3)	-0.0226(2) 0.0963(2)	0.58921 (10)	0.0529	
$\Omega_{3,11}^{3,10} \Omega_{5,10}^{15,18}$ (1) (1) (1) (1) (1) (1) (1) (1) (1) (1)			F29A	1.2366 (3)	0.0994 (3)	0.51805(10) 0.5082(1)	0.0558	
$0^{-1}.0^{-1}.0^{-1}$ Jicos-7-ene (1) (1.38 g, 1.64 mmol, 31%; found			F29 <i>B</i>	1.3980 (4)	-0.0109 (3)	0.5606 (2)	0.1078	
$C 37.3, \Pi 2.0, \Gamma 34.3$	$M = 840; C_{20}$	5H ₁₆ F ₂₄ N ₄ requires	s С 37.2, н	F29C	1.3816 (3)	0.1590 (3)	0.5427 (1)	0.0850
1.9, N 0.4, F 54.3, M	840), m.p. 4	13-414 K, which	was slowly	F30A	1.4009 (4)	0.0766 (3)	0.6474 (1)	0.1074
crystallized from n-h	exane.			F30B F30C	1.1986 (4)	0.1810 (3)	0.6726 (1)	0.0902
				N3	1.2298 (4)	0.0103(3)	0.6865 (1)	0.1031
Crystal data				NII	0.6592 (4)	0.1331(3) 0.1212(3)	0.8129(1)	0.0306
			2	N17	0.9393(4)	0.1593(3)	0.6400(1)	0.0301
$C_{26}H_{16}F_{24}N_4$	L	D _x = 1.883 Mg m [−]	-3	N27	1.0602 (5)	0.1374 (3)	0.5872(1)	0.0373
$M_r = 840.40$	Ν	No $K\alpha$ radiation		Cl	0.8937 (5)	0.1606 (4)	0.7200 (2)	0.0326
Triclinic	,	. = 0.71069 Å		C2	0.8212 (5)	0.0979 (3)	0.7671 (2)	0.0329
$P\overline{1}$	C	Cell parameters fro	m 25	C4	0.8876 (5)	0.1396 (4)	0.8546 (2)	0.0357
a = 9.525 (8) Å		reflections		CS	0.7810 (5)	0.1615 (4)	0.9042 (2)	0.0377
h = 12.936(9) Å	A	$= 0.50 + 10.78^{\circ}$		C0 C7	0.7803 (6)	0.0/46 (5)	0.9567 (2)	0.0596
c = 25.821 (0) Å	U	= 9.39 - 10.76		C8	0.5517 (8)	0.1409 (3)	0.9930 (2)	0.0723
c = 23.631(9) A	μ	$c = 0.2100 \text{ mm}^{-1}$		C9	0.5533 (5)	0.1152(4)	0.9795 (2)	0.0004
$\alpha = 78.16 (4)^{-1}$	1	= 296 K		C10	0.6253 (5)	0.1775 (3)	0.8863 (2)	0.0390
$\beta = 83.79 (4)^{\circ}$	E	llock		C12	0.5594 (5)	0.1359 (4)	0.7990 (2)	0.0364
$\gamma = 72.26 (5)^{\circ}$	0	$.40 \times 0.35 \times 0.33$	5 mm	C13	0.6660 (5)	0.1139 (3)	0.7491 (2)	0.0312
V = 2963 (3) Å ³	C	Colourless		C14	0.6599 (4)	0.1978 (3)	0.6962 (2)	0.0295
Z = 4				C15	0.7477 (5)	0.1216 (3)	0.6582 (2)	0.0293
				C10	0.7998 (5)	0.1617 (3)	0.6006 (2)	0.0298
Data collection				C10	0.9076 (5)	0.0984(3) 0.2617(3)	0.6748 (2)	0.0334
Data conection				C20	0.6819 (6)	0.2017(3)	0.7038 (2)	0.0345
CAD-4 diffractomete	r R	$t_{int} = 0.035$		C21	0.9583 (6)	0.2351(5)	0.9477(2) 0.8407(2)	0.0528
$\omega/2\theta$ scans	θ_1	$max = 24.97^{\circ}$		C22	1.0064 (6)	0.0295 (5)	0.8622 (2)	0.0556
Absorption correction	n: h	$= 0 \rightarrow 7$		C23	0.4407 (5)	0.2477 (4)	0.7906 (2)	0.0458
not applied	k	$= -14 \rightarrow 14$		C24	0.4847 (6)	0.0428 (5)	0.8132 (2)	0.0517
0070 manurad raflac	tions l	$= -20 \times 20$		C25	0.7175 (6)	0.2729 (4)	0.5685 (2)	0.0409
9070 measured rened		$-29 \rightarrow 30$		C26	0.8171 (6)	0.0705 (4)	0.5676 (2)	0.0426
839/ independent ret	lections 5	standard reflection	ns	C28	1.1945 (7)	0.1061 (4)	0.5989 (2)	0.0421
4988 observed reflect	tions	trequency: 2.5 h	L	C30	1.3023 (7)	0.0879(6)	0.5527(2) 0.6509(2)	0.0641
$[I > 2\sigma(I)]$		intensity variation	n: none	F51A	1.4144 (4)	0.2158(2)	0.0009(2) 0.3048(1)	0.0082
				F51 <i>B</i>	1.2682 (3)	0.2099 (2)	0.2495 (1)	0.0642
Refinement				F51C	1.2369 (3)	0.1463 (2)	0.3307 (1)	0.0676
				F52A	1.2188 (3)	0.4268 (2)	0.21972 (10)	0.0539
Refinement on F	Ŵ	eighting scheme l	based on	F52 <i>B</i>	1.3566 (3)	0.4324 (2)	0.27843 (10)	0.0618
R = 0.0501		measured e.s.d.'s		F52C	1.1304 (3)	0.5317 (2)	0.27656 (10)	0.0529
wR = 0.0396	(4	Δ/σ) _{max} = 0.0672		F53R	0.0165(4) 0.7561(3)	0.2719(3) 0.2640(2)	0.3000 (1)	0.0814
S = 2.111	Δ	$n_{max} = 0.46 \text{ e} \text{ Å}^{-1}$	3	F53C	0.8490 (3)	0.1867(2)	0.3560(1)	0.0095
4988 reflections	$\overline{\Lambda}$	$\rho_{\rm max} = -0.34 \mathrm{e}^{\mathrm{A}}$	-3	F54A	0.6540 (3)	0.4725 (3)	0.4115 (1)	0.0754
973 parameters	~	tomia conttoning fo		F54 <i>B</i>	0.5358 (3)	0.4939 (3)	0.3426(1)	0.0784
H-stom parameters n	nt .	forme scattering in		F54C	0.7027 (3)	0.5747 (2)	0.3406(1)	0.0731
rafined	01	from internationa	il ladies	F55A	0.6307 (6)	0.2239 (4)	0.1069 (2)	0.1126
Termed		for X-ray Crystal	lography	F55 <i>C</i>	0.4589 (5)	0.3014 (3)	0.1570 (2)	0.0910
		(1974, Vol. IV)		F564	0.0313(4) 0.6437(4)	0.1887(3) 0.5468(3)	0.1889(2)	0.0909
				F56B	0.4452(3)	0.5109(3)	0.0947(1) 0.1265(1)	0.0614
Table 1 Engetiens	1	1 I		F56C	0.5573 (4)	0.4453 (3)	0.0594 (1)	0.1051
Table 1. Fractiona	ι ατοπις ςο	ordinates and e	quivalent	F55A2	0.5214	0.2874	0.1018	0.0633
isotropic d	isplacement	parameters $(Å^2)$		F55 <i>B</i> 2	0.5245	0.2669	0.1860	0.0633
				F55C2	0.7114	0.1801	0.1423	0.0633
U_{eq}	$= (1/3)\Sigma_i\Sigma_jU$	ija [*] a [*] a _i .a _j .		F39A F50P	0.8965 (5)	0.3243 (4)	-0.0185 (1)	0.1223
r	v	7	11	F59C	1.1150 (5)	0.2260 (3)	-0.006/(1)	0.1259
F21A 1.0426 (3)	0.2328 (3)	0.8795(1)	0.0776	F60A	1 2457 (4)	0 3355 (4)	-0.0310(1)	0.1021
F21B 1.0465 (3)	0.2326 (3)	0.7970(1)	0.0739	F60 <i>B</i>	1.1780 (4)	0.2450 (3)	0.1190 (1)	0.0997
F21C 0.8576 (4)	0.3317 (3)	0.8340(1)	0.0755	F60C	1.1159 (3)	0.4186 (3)	0.1054 (1)	0.0732
F22A 1.0913 (3)	0.0157 (3)	0.9022 (1)	0.0830	N33	1.0215 (4)	0.3390 (3)	0.3016(1)	0.0333
F22B 1.0995 (3)	0.0169 (2)	0.81934 (10)	0.0673	N41	0.9304 (4)	0.3907 (3)	0.3450 (1)	0.0360
F22C 0.9506 (3) F23A 0.2522 (2)	-0.0559 (2)	0.8710(1)	0.0676	N47	0.8268 (4)	0.3434 (3)	0.1281 (1)	0.0446
F23B 0.4986 (3)	0.2303 (2)	0.7511(1)	0.0079	C31	0.8/30(3)	0.3328 (3)	0.0783 (2)	0.0442
F23C 0.3492 (3)	0.2637 (2)	0.83344 (10)	0.0595	C32	0.9340 (5)	0.3263 (4)	0.2120(2) 0.2557(2)	0.0380
F24A 0.4090 (3)	0.0400 (3)	0.7732 (1)	0.0728	C34	1.1724 (5)	0.3408 (3)	0.3076(2)	0.0337
F24B 0.5858 (3)	-0.0551 (3)	0.8226 (1)	0.0644	C35	1.1757 (5)	0.3423 (3)	0.3678 (2)	0.0384
F24C 0.3919 (3)	0.0497 (3)	0.8551(1)	0.0704	C36	1.2064 (6)	0.4387 (4)	0.3886 (2)	0.0520
F25A 0.8043 (3)	0.3091 (2)	0.5299 (1)	0.0742	C37	1.2333 (7)	0.3890 (5)	0.4462 (2)	0.0645
F25D U.0049 (4)	0.2666 (2)	0.5459(1)	0.0930	C38	1.1062 (7)	0.3820 (5)	0.4699 (2)	0.0618
F26A 0.6904 (3)	0.3320 (2)	0.39023 (10)	0.0094	C39	0.9905 (5)	0.4296 (4)	0.4300 (2)	0.0475
	3.5.72(2)	0.00700 (10)	0.0333	040	1.0191 (3)	0.3433 (4)	0.3920 (2)	0.0396

C42	0.7832 (5)	0.3823 (3)	0.3398 (2)	0.0339
C43	0.7755 (5)	0.4034 (3)	0.2780 (2)	0.0358
C44	0.7292 (5)	0.3295 (3)	0.2472 (2)	0.0371
C45	0.6992 (5)	0.4044 (3)	0.1929 (2)	0.0409
C46	0.6681 (6)	0.3682 (4)	0.1426 (2)	0.0417
C48	0.8563 (5)	0.4004 (3)	0.1686 (2)	0.0381
C49	0.8778 (5)	0.2448 (4)	0.2365 (2)	0.0421
C50	1.0518 (6)	0.5181 (4)	0.3949 (2)	0.0542
C51	1.2733 (7)	0.2283 (4)	0.2975 (2)	0.0451
C52	1.2184 (7)	0.4331 (4)	0.2705 (2)	0.0431
C53	0.7511 (7)	0.2752 (4)	0.3698 (2)	0.0483
C54	0.6690 (7)	0.4826 (5)	0.3590 (2)	0.0538
C55	0.6036 (8)	0.2719 (5)	0.1473 (2)	0.0599
C56	0.5771 (7)	0.4685 (5)	0.1052 (2)	0.0623
C58	0.9997 (7)	0.3342 (4)	0.0571 (2)	0.0413
C59	1.0192 (8)	0.3189 (5)	0.0002 (2)	0.0627
C60	1.1342 (7)	0.3359 (5)	0.0812 (2)	0.0617

Table 2. Selected geometric parameters (Å, °)

N3—N11	1.487 (5)	N33—N41	1.482 (5)
N3-C2	1.476 (6)	N33—C32	1.463 (5)
N3-C4	1.458 (6)	N33—C34	1.470 (7)
N11-C10	1.468 (6)	N41-C40	1.470 (6)
N11-C12	1.457 (6)	N41-C42	1.462 (7)
N17—C16	1.462 (7)	N47—C46	1.470 (7)
N17-C18	1.515 (5)	N47-C48	1.491 (7)
C1-C2	1.540 (6)	C31-C32	1.539 (7)
C1-C18	1.524 (7)	C31-C48	1.518 (6)
C1-C19	1.518 (5)	C31-C49	1.505 (7)
C2-C13	1.539 (7)	C32C43	1.543 (7)
C4C5	1.564 (6)	C34—C35	1.563 (6)
C5—C6	1.569 (6)	C35—C36	1.570 (8)
C5-C10	1.543 (7)	C35—C40	1.546 (7)
C6—C7	1.514 (8)	C36—C37	1.512 (7)
C6-C20	1.52(1)	C36—C50	1.531 (7)
C7—C8	1.32(1)	C37—C38	1.315 (9)
C8-C9	1.507 (9)	C38—C39	1.493 (7)
C9-C10	1.543 (7)	C39—C40	1.557 (7)
C9-C20	1.517 (7)	C39-C50	1.530 (8)
C12C13	1.571 (6)	C42—C43	1,570 (6)
C13-C14	1.555 (5)	C43—C44	1.544 (8)
C14-C15	1.533 (6)	C44—C45	1.532 (6)
C14-C19	1.525 (7)	C44—C49	1.541 (6)
C15-C16	1.552 (5)	C45-C46	1.554 (7)
C15—C18	1.550 (7)	C45C48	1.549 (7)
C2-N3-C4	125.7 (3)	C32-N33-C34	124.4 (3)
C10-N11-C12	125.8 (3)	C40—N41—C42	126.2 (3)
N3-C4-C5	103.4 (3)	N33-C34-C35	103.7 (3)
C4-C5-C6	123.7 (3)	C34-C35-C36	122.9 (4
N11-C12-C13	103.5 (3)	N41-C42-C43	101.2 (3

One of the cyclobutane CF₃ groups is disordered and has been treated as comprising a major [F55A, F55B, F55C; 81 (1)%] and a minor [F55A2, F55B2, F55C2; 19 (1)%] conformer, with a total occupancy constrained to be 1. The minor conformer has been treated as a rigid group with isotropic vibrational parameters. Data collection: Enraf-Nonius/CAD-4AJ diffractometer control software. Cell refinement: Enraf-Nonius/CAD-4AJ diffractometer control software. Data reduction: TEXSAN PROCESS (Molecular Structure Corporation, 1985). Program(s) used to solve structure: TEXSAN LS. Molecular graphics: ORTEPII (Johnson, 1976). Software used to prepare material for publication: TEXSAN FINISH. Literature search: CSSR (1984).

References

- Barlow, M. G., Suliman, N. S. & Tipping, A. E. (1992). J. Chem. Soc. Chem. Commun. Submitted.
- Burger, K., Schickaneder, H., Hein, F., Gieren, A., Lamm, V. & Engelhardt, H. (1992). Liebigs Ann. Chem. pp. 845-852.
- CSSR (1984). Crystal Structure Search and Retrieval Instruction Manual. SERC Daresbury Laboratory, Warrington, England.
- Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Molecular Structure Corporation (1985). TEXSAN. TEXRAY Structure Analysis Package. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Sheldrick, G. M. (1986). SHELXS86. Crystallographic Computing 3, edited by G. M. Sheldrick, C. Krüger & R. Goddard, pp. 175-189. Oxford Univ. Press.

Acta Cryst. (1994). C50, 556-558

endo-(1*R**,9*R**,10*R**)-9,10-Dimethoxy-12,15,15-trimethyltricyclo[9.3.1.0^{3,8}]pentadeca-3(8),11-diene-4,13-dione

Yasuyuki Takenaka, Taizo Ono,† Akira Uchida,‡ Yuji Ohashi, Takashi Furukawa, Yoshiaki Horiguchi and Isao Kuwajima

Department of Chemistry, Tokyo Institute of Technology, Meguro-ku, Tokyo 152, Japan

(Received 2 August 1993; accepted 4 October 1993)

Abstract

The X-ray analysis of the title compound, $C_{20}H_{28}O_4$, revealed a chair-boat-like conformation of the eightmembered ring which makes the whole molecule *endo* and the relative stereochemistry C1*R*^{*}, C9*R*^{*} and C10*R*^{*}.

Comment

A single-crystal X-ray analysis has been performed for the title compound (2), which was obtained in the course of a study towards the synthesis of taxusin (3) (a congener of taxane diterpenes). The synthetic process has been reported elsewhere (Horiguchi, Furukawa & Kuwajima, 1992). Tricyclic compound (2) was formed by the intramolecular cyclization of (1) and has the necessary

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 71598 (74 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: HU1061]

[†] Present address: Government Industry Research Institute Nagoya, Kita-ku, Nagoya, Aichi 462, Japan.

[‡] Present address: Department of Biomolecular Sciences, Toho University, Miyama, Funabashi, Chiba 274, Japan.